

A critical survey of detection equipment suitable for installation in aerospace vehicles is presented

Subject Headings *Fire, detection, Detection, of fire*

G. L. Isles

Chleck, D., Cucchiara, O., and Donaghue, T. (Parametrics Incorporated, Waltham, Massachusetts) "Detection Techniques for Hazardous Vapors of Elemental Propellants," *Technical Documentary Report No ASD-TDR-63-294 Contract AF 33(657)-8916, Air Force Systems Command, Wright-Patterson Air Force Base, Ohio (July 1963)*

This report describes an investigation into techniques capable of being developed into simple, direct, inexpensive, field devices for detecting hazardous vapors of hydrogen and fluorine, or fluorine-containing oxidizers. Goals for detection limits were one-fourth or less of the lower explosion limit (LEL) for hydrogen and the threshold limit value (TLV) for fluorine.

The results of an extensive literature search of detection techniques are presented. It is concluded that only indirect methods exist and they do not otherwise meet the program objectives.

Three new systems for fluorine are evaluated as well as another technique of potential dual-gas detection capability.

1 The fluorescence of aluminum morin paper, under ultraviolet excitation, is quenched by hydrogen fluoride. Assuming HF is coexistent with elemental fluorine, this effect leads to a sufficiently sensitive, stable and direct system without further development.

2 A hydroquinone-Krypton-85 clathrate compound releases radioactive Krypton-85 gas in proportion to the oxidation of hydroquinone. Thus, a radioactivity counter can determine the oxidizer concentration. A "sensitized" clathrate of improved characteristics was developed that displayed satisfactory response to 0.05 ppm of fluorine (compared to the TLV of 0.1 ppm for 8 hours exposure), was independent of flow rate, except for very short residence times, and was simple and effective for long times.

3 Chemiluminescence results from oxidation of a rhodamine B-silica gel surface. This light excites a photomultiplier tube signal which is proportional to the fluorine oxidizer flux (i.e., concentration times flow rate) passing the detector surface. Response is less than one second for all fluorine concentrations and adequate for determining TLV for 8 hour intervals.

4 The final technique examined makes use of radioactive Krypton-85 incorporated into a solid, i.e., a kryptonate. Fluorine reacts with the surface layers of kryptonates, causing release of Krypton-85. The effluent radioactive gas is directly proportional to fluorine concentration and is measured by a Geiger counter. The highest sensitivity is obtained using kryptonated pyrographite. Its response is adequate for continuous detection of fluorine in the atmosphere. (Kryptonated boron also is considered satisfactory although less sensitive.)

In adapting kryptonates to hydrogen detection, the decreased radioactivity of

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