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web the amount in moles of pb 3 o 4 to 3 significant figures iii the amount in moles of lead produced iv mass of lead produced to 3 significant figures b 35 61 g of tin reacts with exactly 42 60 g of chlorine cl 2 to form 78 21 g of tin iv chloride  $\text{SnCl}_4$  i calculate the number of moles of tin chlorine and tin chloride a r values

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