

HAROLD LASKE

(1893—1950)

A Biographical Memoir

by

KINGSLEY MARTIN

LONDON
VICTOR GOLLANCZ LTD.
1951



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web use the information in this chart to answer the following questions 1 if i have 125 grams of water at a temperature of 25.0 $^{\circ}\text{C}$ and heat it to 145.0 $^{\circ}\text{C}$ what will the change in enthalpy be 2 sketch the heating curve temperature vs time for the process in 1 3 i have 18.7 grams of an unknown liquid chemical at a temperature of 35.0 $^{\circ}\text{C}$

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web sample problem calorimetry and enthalpy changes in an experiment 25.0 ml of 1.00 M HCl at 25.0 $^{\circ}\text{C}$ is added to 25.0 ml of 1.00 M NaOH at 25.0 $^{\circ}\text{C}$ in a foam cup calorimeter a reaction occurs and the temperature rises to 32.0 $^{\circ}\text{C}$ calculate the enthalpy change in kJ for this reaction assume the densities of the solutions are 1.00 g/ml and that

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web choose 1 answer the final temperature is closer to t_1 than to t_2 a the final temperature is closer to t_1 than to t_2 the final temperature is exactly halfway between t_1 and t_2 b the final temperature is exactly halfway between t_1 and t_2 the final temperature is closer to t_2 than to t_1 c

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web 6 a 218 g sample of steam at 121°C is cooled to ice at 14°C find the change in heat content of the system 7 in going from ice at 34°C to steam at 138°C a sample of water absorbs $1.41 \times 10^5 \text{ J}$ find the mass of the sample answers 1 $2.68 \times 10^4 \text{ J}$ 2 $6.23 \times 10^4 \text{ J}$ 3 $1.71 \times 10^4 \text{ J}$ 4 $1.11 \times 10^5 \text{ J}$ 5 $2.11 \times 10^5 \text{ J}$ 6

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