TABLE I An abbreviated list of the CODATA recommended values of the fundamental constants of physics and chemistry based on the 2014 adjustment.

Quantity	Symbol	Numerical value	Unit	Relative std uncert. u,
speed of light in vacuum	$c, c_0$	299 792 458	$m s^{-1}$	exact
magnetic constant	µ <sub>0</sub>	$4\pi \times 10^{-7}$	N A <sup>-2</sup>	
		$= 12.566370614 \times 10^{-7}$	N A <sup>-2</sup>	exact
electric constant 1/μ <sub>0</sub> c <sup>2</sup>	€0	$8.854187817\times 10^{-12}$	$F m^{-1}$	exact
Newtonian constant of gravitation	G	$6.67408(31) \times 10^{-11}$	$m^3 kg^{-1} s^{-2}$	$4.7 \times 10^{-5}$
Planck constant	h	$6.626070040(81)\times 10^{-34}$	Js	$1.2 \times 10^{-8}$
$h/2\pi$	ħ	$1.054571800(13)\times 10^{-34}$	Js	$1.2 \times 10^{-8}$
elementary charge	é	$1.6021766208(98) \times 10^{-19}$	C	$6.1 \times 10^{-9}$
magnetic flux quantum h/2e	$\Phi_0$	$2.067833831(13)\times 10^{-15}$	Wb	$6.1 \times 10^{-9}$
conductance quantum $2e^2/h$	$G_0$	$7.7480917310(18) \times 10^{-5}$	S	$2.3 \times 10^{-10}$
electron mass	$m_{\pi}$	$9.10938356(11) \times 10^{-31}$	kg	$1.2 \times 10^{-8}$
proton mass	$m_{\rm p}$	$1.672621898(21) \times 10^{-27}$	kg	$1.2 \times 10^{-8}$
proton-electron mass ratio	$m_p/m_e$	1836.152 673 89(17)	800m	$9.5 \times 10^{-11}$
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.2973525664(17) \times 10^{-3}$		$2.3 \times 10^{-10}$
inverse fine-structure constant	$\alpha^{-1}$	137.035 999 139(31)		$2.3 \times 10^{-10}$
Rydberg constant $\alpha^2 m_e c/2h$	$R_{\infty}$	10 973 731.568 508(65)	$m^{-1}$	$5.9 \times 10^{-12}$
Avogadro constant	$N_A, L$	$6.022140857(74) \times 10^{23}$	$\text{mol}^{-1}$	$1.2 \times 10^{-8}$
Faraday constant $N_Ae$	F	96 485.332 89(59)	C mol <sup>-1</sup>	$6.2 \times 10^{-9}$
molar gas constant	R	8.314 4598(48)	$J \text{ mol}^{-1} K^{-1}$	$5.7 \times 10^{-7}$
Boltzmann constant $R/N_A$	k	$1.38064852(79) \times 10^{-23}$	$J K^{-1}$	$5.7 \times 10^{-7}$
Stefan-Boltzmann constant				
$(\pi^2/60)k^4/\hbar^3c^2$	σ	$5.670367(13) \times 10^{-8}$	$W m^{-2} K^{-4}$	$2.3\times10^{-6}$
		coepted for use with the SI		400000000000000000000000000000000000000
electron volt (e/C) J	eV	$1.6021766208(98) \times 10^{-19}$	J	$6.1 \times 10^{-9}$
(unified) atomic mass unit $\frac{1}{12}m(^{12}C)$	u	$1.660539040(20)\times10^{-27}$	kg	$1.2\times10^{-8}$

# **Fundamentals Of Physical Constants**

**Brendan G. Carr** 

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