

## EFFECTIVE SCHOOL LEADERSHIP

- Empowered leader is driving teacher and increasing in turn

## TRANSFORMATIONAL LEADERSHIP

- Transformational leadership is a leadership style that focuses on inspiring and motivating followers to achieve their best work

EDUCATIONAL SUPERVISOR'S LEADERSHIP  
LEADERSHIP LEADERSHIP

## Rejuvenating Transformational

- Rejuvenate the school's vision and mission
- Rejuvenate the school's culture and climate
- Rejuvenate the school's leadership and management
- Rejuvenate the school's curriculum and instruction
- Rejuvenate the school's assessment and evaluation

## SUPPORT DISTRIBUTIONAL LEADERSHIP

- Distributive leadership is a leadership style that focuses on sharing power and decision-making among all members of the organization
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- Distributive leadership is a leadership style that focuses on sharing power and decision-making among all members of the organization

## ENSURE ACADEMIC EXCELLENCE

- Knowledge, skills, and abilities to ensure academic excellence
- Academic excellence is a leadership style that focuses on ensuring that all students achieve high academic achievement

## Educational Educational Leadership

Effective educational leadership is a leadership style that focuses on ensuring that all students achieve high academic achievement

# SCHOOL LEADERSHIP FOR EFFECTIVE SCHOOL MANAGEMENT

## Design School

Design school is a school that is designed to be a model of effective school management

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## TRANSFORMATIONAL LEADERSHIP

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## Administrative

- Administrative leadership is a leadership style that focuses on ensuring that all students achieve high academic achievement
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REQUIREMENT OF FUTURE

CHALLENGE OF SCHOOL LEADERSHIP FOR SCHOOL SUCCESS

IN SERVICE

LEADERSHIP LEADERSHIP

LEADERSHIP LEADERSHIP

- Academic excellence is a leadership style that focuses on ensuring that all students achieve high academic achievement
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Academic excellence is a leadership style that focuses on ensuring that all students achieve high academic achievement

# Effective School Leaders Guide To Management

**Vernon G. Smith**



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web aug 30 2022 acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1  
 $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o l}$   
 one mole of hcl would be fully neutralized by one mole of naoh

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide naoh and oxalic acid  $\text{h}_2\text{c}_2\text{o}_4 \cdot 2\text{h}_2\text{o}$  and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak  $\text{pK}_a$  of an unknown acid or  $\text{pK}_b$  of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction  $\text{NaOH} + \text{C}_6\text{H}_4\text{COOH} \rightarrow \text{C}_6\text{H}_4\text{COONa} + \text{H}_2\text{O}$  the point of equivalence is the point of titration when the number of moles of naoh added is equal to the number of moles of khp in the

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web plants for shade most thriving in full shade or part shade in a variety of soil conditions including dry shade suitable under trees shrubs in the lee of walls fences etc includes many ferns early spring flowers such as anemones snowdrops epimedium and foliage plants like hosta and heuchera many grown in our woodland garden

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