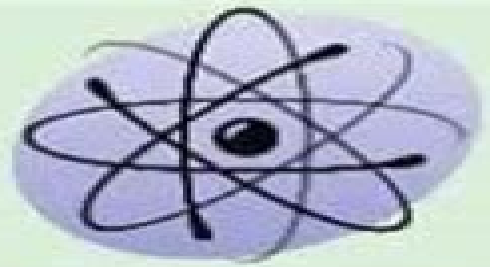


Fundamental Physical Constants



Source: www.physicshelp.com

Fundamental Physical Constants

Quantity - Symbol - Value

Planck constant - h - $6.62 \times 10^{-34} \text{ J}\cdot\text{s}$

Atomic mass constant - u - $1.66 \times 10^{-27} \text{ kg}$

Avogadro's number - N - $6.022 \times 10^{23} \cdot \text{mol}^{-1}$

Boltzmann constant - k - $1.38 \times 10^{-23} \text{ J}\cdot\text{K}^{-1}$

Molar gas constant - R - $8.314 \text{ J}\cdot\text{K}^{-1} \text{ mol}^{-1}$

Speed of light - c - $2.99 \times 10^8 \text{ m / s}$

Charge of electron - e - $1.60 \times 10^{-19} \text{ C}$

Coulomb constant - K - $8.98 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$

Proton mass - m_p - $1.67 \times 10^{-27} \text{ kg}$

Neutron mass - m_n - $1.67 \times 10^{-27} \text{ kg}$

Fundamentals Measures And Constants For Science And Technology

WJ Hussar



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